ISSAI FOOD SAFETY AND STANDARDS	Determination of Vitamin B12 (Cyanocobalamin) in Fortified Rice			
Inspiring Trust, Assuring Safe & Nutritious Food Ministry of Health and Family Welfare, Government of India	Kernel (FRK)			
Method No.	FSSAI.FRK.16.006.2023	Revision No. & Date	0.0	
Scope	This method is applicable for th	e quantitative analysis of	Vitamin B12 as	
	Cyanocobalamin at an LOQ of 25 µg/Kg using Liquid chromatography			
	coupled with Tandem Mass Spectrometer (LC-MS/MS).			
Caution	Methanol is a flammable Liquid. Handle in a hood away from flames			
	Sodium hydroxide is caustic. Contact with very high concentrations of			
	sodium hydroxide can cause severe burns to the eyes, skin, digestive system			
	or lungs. Prolonged or repeated skin contact may cause dermatitis. Handle			
	with care.			
	Formic Acid is a corrosive chemical and contact can severely irritate and			
	burn the skin and eyes with possible eye damage. Inhaling formic acid can			
	irritate the nose and throat. Use in a fume hood			
Principle	Cyanocobalamin is extracted with the diluent (50% Methanol containing			
	0.1% Formic acid) and $\alpha$ -amylase. The extract is then diluted with water,			
	filtered, diluted with diluent and the analysed by LC-MS/MS.			
Apparatus/Instruments	1. LC-MS/MS, System equipped with a Binary gradient pump, an auto			
	sampler and tandem mass spectre	ometer.		
	2. Analytical Balance, -Suitable	for weighing samples wi	th accuracy up to	
	0.0.0001 g			
	3. Centrifuge, -5000 rpm, that can accommodate 50 mL tubes			
	4. Amber colored volumetric flask (25 mL)			
	5. Volumetric flask: 1000 mL			
	6. Measuring cylinder 1000 mL			
	7. Micropipettes capable of delivering from 100 -1000 µl, 20 -200 µl10 -			
	100 µl.			
	8. Shaker incubator			
	9. Column: Kinetex 2.6 µm, XB	C18 Column, $2.1 \times 100$	mm or equivalent	
	10. Sonicator.			
	11. Vortex mixer.			
	12. Homogenizer for sample gr	inding		
Materials and Reagents	1. Ammonium formate, MS	Grade		
	2. Methanol, LR Grade.			
	3. Formic acid, MS Grade.			
	4. Sodium hydroxide, LR G	rade		
	5. α-Amylase, (TCI, A0447	) or equivalent		
	6. CRM Cyanocobalamin (C	CAS No: 68199, P.No: V2	2876, Sigma	
<b>D</b>	Aldrich) or equivalent			
Preparation of Reagents	a) Mobile phase A (5 mM Am	monium formate in wat	er)	
	1. Weigh accurately 0.315 g	g of Ammonium formate.		

	2. Transfer into a 1000 mL of volumetric flask.	
	3. Add Milli-Q Water to dissolve and make-up to 1000 mL.	
	4. Sonicate for 15 mins.	
	5. Filter through 0.45 µm filter.	
	b) Mobile phase B (100% Methanol)	
	Transfer 1000 mL Methanol to mobile phase glass reservoir and sonicate	
	for 15 min.	
	c) Diluent (50% Methanol containing 0.1 % Formic acid)	
	1. Transfer 500 mL Methanol into 1000 mL measuring cylinder.	
	2. Add 1 mL Formic acid.	
	3. Add water up to mark 1000 mL.	
	4. Mix well and sonicate for 15 min.	
Sample Preparation	1.Grind 50 g of FRK into a fine powder.	
	2. Accurately weigh 5 g ( $\pm$ 0.5 g) of ground sample into a 25 mL amber	
	colored volumetric Flask	
	3.Add 50 mg $\alpha$ -amylase and 20 mL of diluent	
	4.Vortex for 5 min.	
	5.Make-up the volume to 25 mL using diluent.	
	6.Sonicate for 20 min.	
	7.Allow the sample to come to room temperature (25 $^{\circ}$ C).	
	8. Filter the sample using a syringe filter it (0.45 $\mu$ m).	
	9.Use the filtrate for LC-MS/MS analysis.	
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	Preparation of intermediate standard solution – 3 (ISS-3) (1 mg/kg)					
	1. Pipette out 1.0 mL of ISS-2.					
	2. Transfer to a a 10 mL amber colored volumetric flask 2 mL of Milli Q					
	Water.					
	3. Add diluent and make-up to 10 mL.					
	4. Vortex for 2 min.					
	Preparation of calibration standard solutions					
	Use ISS – 3 (1 mg/kg) for preparing Calibration standard solution as					
	indicated Table below.					
		ISS - 3	V OI OI		Final	Final
	Standard	(1 mg/kg))	1SS - 3	diluent	vol.	conc.
	1.07	1000	(mL)	(mL)	(mL)	(µg/Kg)
	LS7	1000	2.000	8.000	10	200
	LS6	1000	1.000	9.000	10	100
	LS5	1000	0.500	9.500	10	50
	LS4	1000	0.200	9.800	10	20
	LS3	1000	0.100	9.900	10	10
	LS2	1000	0.050	9.950	10	5.0
	LS1	1000	0.025	9.975	10	2.5
	NOTE: Alway	s use freshly <sub>l</sub>	prepared ca	libration st	andards	
Chromatographic	a) Instrument: LC-MS/MS Spectrometer.					
Conditions	b) Chromatographic Conditions: As detailed in below Table					
	Instrument		LC-MS/M	IS		
	Detector		Mass Detector			
	Column		Kinetex (2.6µm, XB C18 Column, 2.1 x 100			
			mm)			
	Run time		7 min			
	Column Tem	perature	35°C			
	Flow rate		0.25 mL/1	nin		
	Injection Vol	ume	20 µl			
	Mobile Phase	e A	5 mM An	nmonium fo	rmate	
	Mobile Phase	e B	Methanol			
	Diluent		50% Metl	nanol contai	ning 0.1 %	Formic acid
	Source Temp	erature	140 °C			
	Desolvation Temperature 300 °C					
	MRM (Quant	tifier)	678>147			
	MRM (Qualit	fier)	678>359			
	CE         26 V           CV         35 V					
	Source ESI + VE					
			·			

	c) LC-Grad	lient Program		
	Time (min	Flow rate	A (%)	( <b>B</b> )%
		(mL/min)	A (70)	
	0.00	0.25	90	10
	2.00	0.25	90	10
	4.00	0.25	10	90
	5.00	0.25	90	10
	7.00		90 FLC MS/MS 3	10
	Note: The laboratory may use any model of LC-MS/MS instrument after appropriate tuning and optimization. Instrument tuning and settings vary with make and model. Set parameter as per manufacturer's			
	instructions and optimize the method to achieve the desired LOD and LOQ			
Chromatographic	The sequence of analysis is listed below			
	SL.NO.	NAME OF INJECTIO	NS N IN	UMBER OF NJECTIONS
	1	Blank		2
	2	Linearity Solution (LS) - 1		1
	3	Linearity Solution (LS) - 2		1
	4	Linearity Solution (LS) - 3		1
	5	Linearity Solution (LS) - 4		1
	6	Linearity Solution (LS) - 5		1
	7	Linearity Solution (LS) - 6		1
	8	Linearity Solution (LS) - 7		1
	9	Blank		2
	10	Sample Solution		1
	11	Blank		2
	12	Spike Sample Solution		1
		TOTAL INJECT	IONS	15
Calculation with units of	a) Construct a calibration curve and carry out a regression analysis. by			
Expression	fitting the data into a linear regression curve, including zero as the			
	response for the reagent blank. The Regression coefficient ( $\mathbb{R}^2$ ) of			
	should be >0.99			
	b) Calculate the concentration of Cyanocobalamin using the formula $U_{\alpha} = C \times V_{\alpha}$			
	Where $Cyanocobalamine(\frac{\mu g}{kg}) = \frac{C \times V}{W}$			
	C= concentrat	ion cyanocobalamine in samp	ple	
	V=Make-up v	olume		
	W= Mass of sample taken in g			
	c) The LOD a	and LOQ are determined by c	onsidering the	S/N of 3 and 10,

	respectively, for the Cyanocobalamin (Vitamin B12) signal in the matrix. d) Determine the recovery of Cyanocobalamin (Vitamin B12) at spike level (50 µg/Kg) in sample in six replicates. Calculate the recovery value using the following equation: $Recovery(\%) = \frac{(A - B)}{C} \times 100$ Where: A = the concentration of Vitamin B12 in the spiked sample (µg/kg) B = the content of Vitamin B12 in the control sample (µg/kg)		
	$C =$ the spiked concentration of Vitamin B12 ( $\mu g/kg$ )		
A representative	EFRAC_03082022_81 Vitamin B12 Vitamin B12 072.2012 593.175 3.84 678.207 > 593.175		
chromatogram	95 5 5 025 050 075 100 125 150 175 200 225 250 275 300 325 350 375 400 425 450 475 500 525 650 675 600 625 650 675		
Reference	Method Protocol: PRT/RA/FRK/2022/004, Method Validation Report for Estimation of Cyanocobalamin (Vitamin B12) in Fortified Rice Kernel by LC-MS/MS.		
	AOAC 2011.10 – Single Laboratory Validation of AOAC Official method 2011.10 for Vitamin B12 in Indian infant and Pediatric formulas and Adult Nutritionals.		
Approved by	Scientific Panel on Methods of Sampling and Analysis		